

POLYAK, G.L.; ADRIANOV, V.N.

Algebra of resolvent fluxes in radiant exchange. Inzh.-fiz. zhur. 5
no.7:70-77 J1 '62. (MIRA 15:7)

1. Energeticheskiy institut imeni G.M.Krzhizhanovskogo, Moskva.
(Heat—Radiation and absorption)

ACCESSION NR: AP4038664

S/0170/64/000/004/0074/0080

AUTHOR: Adrianov, V. N.; Polyak, G. L.

TITLE: Differential methods of studying radiative heat transfer

SOURCE: Inzhenerno-fizicheskii zhurnal, no. 4, 1964, 74-80

TOPIC TAGS: Radiative heat transfer, heat exchange, heat radiation

ABSTRACT: The article reviews differential methods of studying radiative heat transfer which because of their relative simplicity have opened up new possibilities. The development of these methods is presented in chronological order, the names of the originators are given, and the methods are compared. Because the differential methods are based on approximate differential equations of heat radiation, they have undergone constant refinement, and this appears to be the direction in which they will continue to develop in the future. Orig. art. has: 16 formulas.

ASSOCIATION: Energeticheskii institut im. G. M. Krzhizhanovskogo, Moscow
(Institute of Power Engineering)

Card 1/2

ACCESSION NR: AP4041072

S/0170/64/000/006/0063/0069

AUTHOR: Polyak, G. L.

TITLE: . New method of investigating radiative heat transfer

SOURCE: Inzhenerno-fizicheskiy zhurnal, no. 6, 1964, 63-69

TOPIC TAGS: heat transfer, radiative heat transfer

ABSTRACT: The method proposed is based on the solution of differential equations in which the tensor character of the radiation field is accounted for. Calculations for a plane layer carried out by this method are of high accuracy, as is demonstrated with an example of the solution of a one-dimensional problem. Also mentioned are works of other authors in this field such as S. N. Shorin, R. K. Kopakov, H. S. Hottel, Van der Held, etc. Orig. art. has: 2 figures and 23 formulas.

ASSOCIATION: Energeticheskii institut im. G. M. Krzhizhanovskogo, Moscow
(Power Engineering Institute)

Card 1/2

20

POLYAK, G. L.

Theory of Heat Exchange in Combustion Chambers.
(In Russian.) G. L. Polyak and S. N. Shorin. *Izvestiya Akademii Nauk SSSR* (Bulletin of the Academy of Sciences of the USSR), Section of Technical Sciences, Dec. 1949, p. 1832-1847.

Proposes a formula for calculation of the above. Principles of this new method are thoroughly analyzed. Experimental investigations confirmed validity of the formulas.

Energetics Inst.-im. Krzhyzhenovskiy

ASIS-SLA METALLURGICAL LITERATURE CLASSIFICATION

POLYAK, G.L., inzh.; SALITA, P.Z., inzh.

Use of d.c. transmission to improve stability. Elektrichestvo no.
5: 12-14 My '58. (MIRA 11:7)

1. Nauchno-issledovatel'skiy institut postoyannogo toka.
(Electric power distribution--Direct current)

24(8)

PHASE I BOOK EXPLOITATION

SOV/1826

Akademiya nauk SSSR. Energeticheskii institut

Teplotoobmena i teplovoe modelirovaniye (Heat Transfer and Modeling of Heat Processes). Moscow, Izd-vo AN SSSR, 1959. 419 p. Errata slip inserted. 3,500 copies printed.

Reep. Ed.: M. A. Mikheyev, Academician; Ed. of Publishing House: D. A. Ivanova; Tech. Ed.: G. M. Shvachko.

PURPOSE: The book is intended for scientists concerned with heat transfer, heat exchanger, and hydraulics of liquid metals, etc.

COVERAGE: This collection is dedicated to the memory of Academician M. V. Karpichev who in the twenties initiated a systematic investigation of heat transfer processes and the efficiency of heat exchangers. Later he led the development of research work in this field. Two scientific collections devoted to works of Karpichev's school have been published, one in 1936, Materialy soveshchaniya po modelirovaniyu (Materials of the Conference on Modeling) and in 1951, Teoriya podobiya i modelirovaniya (Theory of Similitude and Modeling). The present collection prepared in 1956 represents further development of the work of this school. This theory is fundamental for the analysis of many heat problems in the field of electrical and radio engineering. Of great importance are the first systematic investigations of heat transfer and the hydraulics of liquid metals which are of great interest for the result of special investigations of some cases of engineering. As a factor, a dependence of the process on the kind of liquid, temperature, pressure, direction of the flow, and of the wide for realization of experimental data, new dependable recommendations for heat transfer and engineering equipment were developed. Of no less interest is the work on heat transmission in boiling liquids and the condensation of vapors. All investigations are based on the theory of similitude, the nature of which, according to M. V. Karpichev, is that of "self-similarity". Work on the theory of a regular regime applied to a wide class of bodies with an internal source of heat is of interest for the future.

Shorin, S. N.; G. L. Polyski; I. P. Kolchenova; V. N. Arlanov, and U. N. Yermolov. The article gives fundamentals of the theory of radiation heat transfer 165

exchange in transparent and in illuminated media. It describes sources of light and changes of illumination and gives a photographic method for measuring streams of light. Investigations of radiation exchanges in cylindrical chambers and in banks of pipes, transfer of radiation energy in illuminated medium, local illumination of walls of boiler burners and hearth bottoms of open hearth steel furnaces are described. The following personalities are mentioned: O. Ye. Vlasov (approximate solution of a cylindrical equation), Ye. S. Kuznetsov (determination of radiation heat transfer in light models), L. A. Valls (light modeling), G. L. Polyski, and S. M. Shorin (theory of radiation exchanges), and with the cooperation of V. I. Kondratyev (radiation exchanges in banks of pipes). The section on photographic method of measuring "light streams" was compiled by G. L. Polyski; the section "Investigation of the transfer of Radiation Energy in an illuminated medium" was compiled by I. P. Kolchenova, S. M. Shorin and V. N. Arlanov; sections on Measurement of Local Illumination of Walls of Boiler Burners and Investigation of Local Illumination of the Surface of the Bottom of an Open-hearth Furnace" were compiled by O. N. Yermolov and S. N. Shorin. There are 27 references: 19 Soviet, 5 English and 3 German.

AVAILABLE: Library of Congress

Card 20/20

14/4th
8-7-59

ADRIANOV, V.N.; POLYAK, G.L.

Differential methods for radiative heat transfer study. Inzh.-fiz.
zhur. 7 no.4:74-80 Ap '64. (MIRA 17:4)

1. Energeticheskiy institut imeni G.M.Krzhizhanovskogo, Moskva.

PAGE 1 BOOK INFORMATION 807/3407

Abdumirza Mest 8888. Energeticheskiy Institut im. G.M. Krzhizhakovskogo
Problemy energetiki: sbornik nauchnykh i tekhnicheskikh statei posvyashchennykh 50-letiyu G.M. Krzhizhakovskogo
(Problems of Power Engineering: Collection of Articles Dedicated to the
Anniversary of G.M. Krzhizhovsky) Moscow, 1959. 651 p. Errata ally inserted.
2,500 copies printed.

Ms. of Publishing House: S.M. Astrakhan, P.V. Dubrov, P.I. Dubrov, and
S.M. Myrskov; Tech. Ed.: G.A. Prusakov; Editorial Board: A.V. Vinter,
Academician (honored), V.I. Popov (Resp. Ed.) Corresponding Member,
Academy of Sciences USSR, V.I. Vozes, A.S. Fedorovskiy, K.A. Pyrkovich,
S.P. Chudakov, S.B. Bogdanov, Candidate of Technical Sciences, S.I. Kozlov,
Candidate of Technical Sciences, S.M. Lebedev, Candidate of Technical Sciences,
and I.B. Smolov.

PREFACE: This collection of articles is intended as a tribute to the memory
of Academician G.M. Krzhizhovsky.

CONTENTS: The collection contains sixty articles by former students and
colleagues of the deceased Academician. The articles deal with problems
of a wide range of subjects in the field of power engineering: problems
of the regional development of electrical and thermal power engineering,
power engineering technology and the physics of combustion. No personalities
are mentioned. References are given after most articles.

Bukov, Yu.G., I.A. Silyov. Investigation of Heat Exchange in
Molecular Condensation of Pure Vapors 411

Burakov, Yu.A. Basic Methods of the Present Theory of Heat Exchange
of Radiation 423

Aleksandrov, V.S., G.M. Vysok. Photographic Method of Measuring Intensive
Flames 470

Syrovatkin, M.A., I. Kh. Kozlovskiy, and L.K. Drobokov. Effect of
the Rules of Solubility of Substances in Water Vapor on Boiler
Water 483

Petrov, Ye.M. The Role of Science in the Development of Soviet Mind
Technology 496

Syrovatkin, M.A., M.S. Shklyar. Results of the Activity of the
Commission for High Parameter Steam and Scientific Tasks in
Increasing the Reliability and Economy of Thermal Electric Power
Stations in the Future 506

Chudakov, S.P. Basic Principles of Power Engineering 543

Chudakov, S.P. Problem of the Mechanism of Thermal Decomposition
of Fuels 564

Shapovalov, Ye.A. Dynamics of the Process of Separating Volatile
Substances from Solid Fuels 575

Kalashnikov, V.V. High-Speed "Termination" of Solid Fuels (Retarded
Combustion) 583

Kashurichov, A.P. Intensity of Heating Fuels and Control of the
Process of Their Thermal Decomposition 593

Kultrish, L.N. Theory of Combustion and Problems of Intensification
of the Processes of Burning 605

Spyrider, V.A., V.N. Iyevlev, V.I. Anisimov, B.B. Gaimov. Burning
of Turbulent Gas-Liquid Systems in Wallflow Fireproof Chambers 637

Shchegolev, Yu.P., V.G. Yevlev. Two-Stage High-Speed Furnaces 659

Lytev, A.Y. Mass-Heat Exchange in State and Chemical Transformations 675

Saidov, M.S. Heating Deep Substances 681

Chudakov, S.P., A.M. Shchegolev, A.P. Kashurichov. Utilization of Out
-put in Power Engineering 687

Solovtsov, B.I. Flow of Gas During Ignition Occurring Beyond the
Shock Wave 733

Pavlov, V.S. Structure of Heterogeneous Flows in a Shock Front 745

Prevalent, A.S. Motion of Combustion Zone as a Hydrodynamic
Instability 793

Detsenko, B.B. Making Suberland Formulae More Precise for Kinetic
-Gas Coefficients 817

Prevalent, A.P. Physical and Chemical Properties of Thermistors
Manufactured from Monoclinic Oxide 823

S/081/61/000/024/034/086
B117/B147

245200

AUTHOR: Polyak, G. L.

TITLE: Radiant heat exchange of bodies with arbitrary indicatrices of surface reflection

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 24, 1961, 287, abstract 24I71 (Sb. "Konvektivn. i luchistyy teploobmen", M., AN SSSR, 1960, 113 - 132)

TEXT: The description of radiant exchange of bodies requires the use of quantities characterizing the spatial non-uniformity of radiation and reflection. The author develops a theory of radiant exchange, and introduces an integral equation for the brightness of radiation in a transparent medium. The equation was derived for the general case of radiant heat exchange in an arbitrarily closed volume when considering the radiation at the given point M, and is valid for monochromatic radiation. Equations are presented for systems with ideally scattered reflection from both surfaces as well as for the case where the ideally scattering

Card 1/2

Radiant heat exchange ...

S/081/61/000/024/034/086

B117/B147

surface is the envelope and where the internal one has an arbitrary reflection indicatrix. Furthermore, the case was considered, in which the nature of radiation currents is significantly changed only by one change of the reflection indicatrix from one surface (parabolic reflector with a white and a mirror surface at total reflection). The field of current lines of the radiation vector and the fields of illumination are graphically represented for this case. For the total radiation it is recommended that a brightness factor averaged over the spectrum should be introduced, whereby the integration of the initial equation can be simplified. ✓

[Abstracter's note: Complete translation.]

Card 2/2

ADRIANOV, V.N.; POLYAK, G.L.

Using the photographic method for the light modeling of radiant heat exchange. Zhur.nauch.i prikl.fot.i kin. 5 no.2:123-132 Mr-Ap '60.

(MIRA 14:5)

1. Energeticheskiy institut AN SSSR.
(Photographic sensitometry)
(Heat—Radiation and absorption)

ACC NR: AT7003571

(N)

SOURCE CODE: UR/0000/66/000/000/0281/0290

AUTHORS: Dorri, M. Kh.; Polyak, G. L.

ORG: Institute of Automation and Remote Control, Moscow (Institut avtomatiki i telemekhaniki)

TITLE: Mathematical modeling of heat exchangers containing internal heat sources with variable heat-transfer agent consumption

SOURCE: AN BSSR. Institut teplo- i massoobmena. Issledovaniye teplo- i massoobmena v tekhnologicheskikh protsessakh i apparatakh (Study of heat and mass transfer in technological processes and apparatus). Minsk, Izd-vo Nauka i tekhnika, 1966, 281-290

TOPIC TAGS: heat transfer, heat exchanger, mathematic model, Laplace equation, frequency characteristic, temperature

ABSTRACT: An analog of a heat exchanger consisting of a cylindrical channel, in which the heat-transfer agent flows at a variable rate, is examined. The equations of this physical model are:

$$a_1 \frac{\partial \theta(x, \tau)}{\partial \tau} = P(\tau) - \alpha(\tau) R [\theta(x, \tau) - t(x, \tau)],$$

$$a_2 \frac{\partial t(x, \tau)}{\partial \tau} + c_2 G(\tau) \frac{\partial t(x, \tau)}{\partial x} = \alpha(\tau) R [\theta(x, \tau) - t(x, \tau)],$$

$$a_1 = c_1 \gamma_1 F_1; \quad a_2 = c_2 \gamma_2 F_2.$$

Card 1/3

ACC NR: AT7003571

The problem consists of determining the output temperature of the heat-transfer agent $t_2(\tau) = t(L, \tau)$ as a function of the input temperature $t_1(\tau) = t(0, \tau)$, the heat release in the wall $P(\tau)$, and the flow rate of the heat-transfer agent $G(\tau)$. A closed system of equations for modeling the heat exchanger is obtained:

$$b_s \frac{d\bar{\theta}(\tau)}{d\tau} = b_s P(\tau) - b_1 u(\tau) [\bar{\theta}(\tau) - \bar{t}(\tau)];$$

$$\frac{d\bar{t}}{d\tau} + G[t_2(\tau) - t_1(\tau)] = b_1 u(\tau) [\bar{\theta}(\tau) - \bar{t}(\tau)];$$

$$\frac{dt_2}{d\tau} = 12G(\tau)\bar{t} - 6G(\tau)[t_2(\tau) + t_1(\tau)] + \frac{dt_1}{d\tau},$$

and its diagram is shown in Fig. 1.

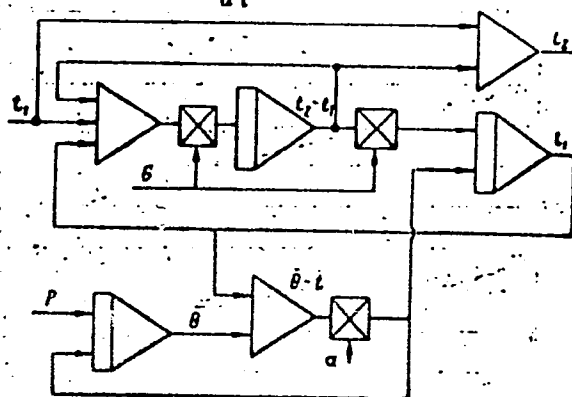


Fig. 1. Diagram of mathematical model of heat exchanger

Card 2/3

Card 3/3

REF ID: A67003062

SOURCE CODE: UR/0103/66/000/008/0116/0123

AUTHOR: Polyak, G. L. (Moscow)

ORG: none

TITLE: Application of the predicting method to some nonlinear systems

SOURCE: Avtomatika i telemekhanika, no. 8, 1966, 116-123

TOPIC TAGS: nonlinear control system, optimal control

ABSTRACT: A basis is given for optimal control with prediction for objects consisting of series-connected first order links, linear or with monotonous characteristics plus second order links which are subjected to conditions similar to the conditions of non-oscillation for linear systems. The method is based on having a computer perform accelerated calculation of future optimal trajectories from a point corresponding to the state of the system at the present moment. In this work, the predictions method is expanded to include systems where curtailment of order is not applicable. Considering the limitations applied to the problem, in most cases optimal control can be performed by the "acceleration - deceleration" principle. The sign of the control in the first interval can usually be determined by observation of the disagreement sign. In other cases two controls must be applied in the first interval, both maximizing and minimizing. Orig. art. has: 2 figures and 25 formulas. [JPRS: 38,836]

SUB CODE: 13 / SUM DATE: 22Nov65 / ORIG REF: 006

Card 1/1 JB

UDC: 62-505.385

0925 2000

L 26395-66 EPF(n)-2/EWT(1)/ETC(f)/EWG(m) WW

ACC NR: AP6007195

SOURCE CODE: UR/0170/66/010/002/0264/0267

AUTHORS: Adrianov, V. N.; Polyak, G. L.

ORG: Moscow Power Institute imeni G. M. Krzhizhanovskiy (Energeticheskiy institut)

TITLE: On the differential method for investigating radiative heat transfer

SOURCE: Inzhenerno-fizicheskiy zhurnal, v. 10, no. 2, 1966, 264-267

TOPIC TAGS: radiative heat transfer, optic thickness, integral equation

ABSTRACT: This article is an answer to P. K. Konakov (whose letter was published in IFZh, 8, No. 3, 1965) who criticized the authors' previous publication (IFZh, 7, No. 4, 1964). Three points brought out by Konakov are refuted. First, according to Konakov, the formulation of boundary conditions relating the radiation flux on the wall q to the wall temperature according to the diffusion method is wrong. The authors show that this method has been generally accepted throughout the world as a proper technique and that Konakov's approach can lead to the erroneous conclusion that the equation

$$(cU)/4 = \sigma_0 T_w^4$$

follows from conditions of local thermodynamic equilibrium. Second, the authors

Card 1/2

UDC: 536.3

L 26395-66

ACC NR: AP6007193

show that Konakov uses Bugar's law but ignores radiation interaction with the medium over a photon mean free path. Third, Konakov claims that the radiation diffusion coefficient equals $c/4k$ instead of $c/3k$. The authors show that this is true only for optically thin gases where $k\delta \ll 1$. Finally, Konakov's attack on Hottel's mathematical analysis as "not clear" is refuted as unfounded. Orig. art. has: 3 formulas and 1 figure.

SUB CODE: 20/ SUBM DATE: 31Jul65/ ORIG REF: 010/ OTH REF: 006

Card 2/2 CC

POLYAK, G.I.; ADRIANOV, V.M.

New method for studying heat transfer by radiation. Inzh. Fiz.
zhur. 7 no.6:63-69 '64. (MIRA 17:12)

1. Energeticheskiy Institut imeni G.M. Krzhizhanovskogo Moskva.

STARODUBTSEV, S. V.; GENERALOVA, V. V.; POLYAK, G. V.

Effect of irradiation conditions on the radiolysis of carbohydrate solutions. Izv. AN Uz.SSR. Ser. fiz.-mat. nauk 7 no.1: 39-45 '63. (MIRA 16:4)

1. Institut yadernoy fiziki AN UzSSR.

(Carbohydrates) (Radiation--Dosage)

POLYAK, G.V.

S/166/63/000/001/004/010
B104/B186

AUTHORS: Starodubtsev, S. V., Generalova, V. V., Polyak, G. V.

TITLE: The influence of the irradiation conditions on the radiolysis of carbohydrate solutions

PERIODICAL: Akademiya nauk Uzbekskoy SSR. Izvestiya. Seriya fiziko-matematicheskikh nauk, no. 1, 1963, 39 - 45

TEXT: Glucose, maltose and saccharose solutions have been irradiated in closed and open ampuls of molybdenum glass by Co^{60} with an activity of $120 \cdot 10^{13}$ g.equ. Ra, the dose rates being varied between 27 and 600 r/sec at temperatures between 0 and 80°C. The aim was to study the influence of the dose rate, the temperature and occluded gases on the properties of this solutions. Results: The rotation of the polarization plane increases with the dose rate. The variation of the specific rotation is the greater the smaller concentration. The polarization plane rotation of a solution depends only slightly on the dose rate, on the irradiation temperature, on the outer pressure and on the existence of occluded gases. The absorption maxima are in the near UV (264 - 270 mμ) and depend linearly on the dose

Card 1/2

S/166/63/000/001/004/010
B104/B186

The influence of the irradiation....

rate in wide range. Dose rate, pressure, presence of oxygen show almost no effect on the absorption maximum. The dioxyacetone yield increases considerably with the irradiation temperature. The upper limit of determination of the absorbed dose depends highly on the concentration of the solutions. At a glucose concentration of 9 % the upper limit is $300 \cdot 10^9$ r, 18 % - $450 \cdot 10^6$ r, 45 % - $800 \cdot 10^6$ r. Similarly, at high dose rates, a new acidic polymer with a molecular weight of 1200 - 4000 and an empirical formula $(C_6H_{10}O_{6.8})_n$ was discovered by S. A. Barker et al., Rad. Res., 16, N3, 1962. There are 6 figures.

ASSOCIATION: Institut yadernoy fiziki AN UzSSR (Institute of Nuclear Physics AS UzSSR)

SUBMITTED: October 9, 1962

Card 2/2

POLYAK, I.

130-7-6/24

AUTHORS: Polyak, I. and Sokolov, O.

TITLE: All-Union Conference of Steel Melters. (Vsesoyuznoye Soveshchaniye staleplavil'shchikov)

PERIODICAL: Metallurg, 1957, Nr 7, pp.11-12 (USSR)

ABSTRACT: Most of this account of a conference of steel melters held in Sverdlovsk in May, 1957, is devoted to a long report by A.F. Martsymov dealing with the main lines of development of steel-making practice and problems involved in further increases of steel-production rates. After indicating recent improvements in Soviet practice and productivity he stated that recent trends in open-hearth (producing 90% of all steel in the USSR), mentioning the 500-ton furnaces in the imeni Voroshilova works and the production rates of 32-34 tons of steel per hour achieved in the 370-ton furnaces at the Kuznetsk and Magnitogorsk metallurgical combines. Dealing in detail with the use of oxygen, Martsymov said that in 1956 20% of all open-hearth steel was produced using oxygen and that experiments on pre-refining (desiliconization) had been successful at several works. Basic refractories had contributed to the increased production rates and all Soviet furnaces would be given basic roofs during 1957-1958. The use

Card 1/2

POLYAK, I.

Fourth Plenum of the Scientific Technological Society of Ferrous
and Nonferrous Metallurgy. Metallurg 6 no.4:40 Ap '61.

(MIRA 14:3)

(Metallurgy--Congresses)

POLYAK, I.B., inzh.

High-caloric fuel for open-hearth furnaces. Metallurg 7
no.2:26-27 F '62. (MIRA 15:3)
(Open-hearth furnaces--Design and construction)
(Gas, Natural)

KOCHINEV, Ye.V., spetsred.; NOVIKOVA, F.M., spetsred.; POLYAK, I.B.,
spetsred.; MAKAROVA, E.A., red.; SHAIRINA, N.D., tekhn.red.

[Metallurgists are fighting for technical progress] Metallurgi
v bor'be za tekhnicheskii progress. Moskva, Izd-vo VTsSPS
Profizdat, 1959. 52 p. (MIRA 13:4)
(Metallurgy) (Rolling (Metalwork))

OYKS, Grigoriy Naumovich, doktor tekhn.nauk; POLYAK, I.B., red.; SUKHAREVA,
R.A., tekhn.red.

[Using oxygen in open-hearth steelmaking] Primenenie kisloroda
v martenovskom proizvodstve stali. Moskva, 1958. 67 p. (Steno-
gramma leksii. Seriya "Metallurgiya," nos.5/6) (MIRA 12:4)
(Open-hearth process) (Oxygen--Industrial applications)

POLYAK, Isaak Berkovich; TRET'YAKOV, Yevgeniy Vasil'yevich;
LANOVSKAYA, M.R., red. izd-va; MIKHAYLOVA, V.V., tekhn.
red.

[Open-hearth production of steel] Martenovskoe proizvodstvo
stali. Moskva, Metallurgizdat, 1963. 161 p. (MIRA 16:6)
(Steel--Metallurgy) (Open-hearth process)

POLYAK, I.G.

Some economic problems concerning the cultivation of
Polygonum coriarium Grig. Vop. biol. i kraev. med. no.4:
201-205 '63. (MIRA 17:2)

21653

S/109/61/006/003/008/018
E052/E314

26.2/31

AUTHOR: Polyak, I.M.

TITLE: Electron Currents in Penning-type Tubes

PERIODICAL: Radiotekhnika i elektronika, 1961, Vol. 6, No. 3,
pp. 395 - 398

TEXT: The present author investigates the conditions under which an electron current may replace the ion current in the cathode circuit of a Penning gauge. The Penning tube has been widely used as an ionisation gauge, an ion source and a high-vacuum pump. High-frequency oscillations have been found to occur in these tubes. The mechanism responsible for the processes occurring in Penning tubes has been investigated by Reykhrudel' et al in Refs. 9 and 10, who were concerned with the ion current under normal working conditions of the device. In the present experiments it was established that there exists a definite value of the magnetic field-strength H at a given anode voltage U_a for which an electron current rather than an ion current will flow in the cathode circuit. The experimental tube employed is shown in Fig. 1.

Card 1/8

21653

S/109/61/006/003/008/018

EC32/E314

Electron Currents in

The cathode was in the form of a Wehnelt cylinder 1, with a small aperture at one end, as shown. A heated tungsten spiral was placed in the plane of this aperture. The anode 2 and the plane collector 3 were of the usual form. The tube was placed in a uniform magnetic field produced by a long solenoid. Arrangements were made for the displacement of the electrodes, under vacuum, when necessary. In the present experiments, the magnetic field was directed along the z-axis (Fig. 1). Smirnitskaya and Reykhrudel' (Ref. 11) have calculated the form of the trajectories of the electrons in the Penning tube. It was found that for the geometry illustrated in Fig. 1, the equations of motion for an electron emitted from a point $(-d, 0)$ are of the form:

$$\ddot{z} + \frac{2eU_0}{md^2} z = 0, \quad (3)$$

$$\dot{\theta} = \frac{eH}{2mC}, \quad (4)$$

Card 2/8

$$\ddot{r} + \left[\left(\frac{eH}{2mC} \right)^2 - \frac{eU_0}{md^2} \right] r = 0. \quad (5)$$

21653

S/109/61/006/003/008/018
E032/E314

Electron Currents in

The latter equation has three solutions, namely,
 $r = A \sin h(\omega t + \alpha)$ when $H < H_c$, $r = Bt + C$ when
 $H = H_c$ and $r = D \sin(\omega t + \beta)$ when $H > H_c$. H_c is the
critical magnetic field given by

$$\frac{eH_c}{2mC} = \left(\frac{eU_o}{md^2} \right)^{1/2} \quad (6) .$$

The present author has measured the collector current I_3 as
a function of the magnetic field H for different values of
the anode voltage U_a and the inter-electrode distance d .
A typical curve is shown in Fig. 2 ($d = 15$ mm, $U_a = 440$ V,
 $P = 4.4 \times 10^{-5}$ mm Hg). The electron current is plotted above
the H-axis and the ion current below this axis. For a given
 d , each value of U_a has an associated value of H for

Card 3/8

21653

S/109/61/006/003/008/018
E032/E314

Electron Currents in

which the electron current at the collector reaches a maximum. These values of U_a and H are given in the first two columns of the table for $d = 15$ mm. The third column gives the critical values H_c calculated from Eqs. (2) and (6), subject to the conditions $d = R$, $U(r,z) = U_a$ at $z = 0$ and $r = R$ (R is the radius of the ring anode and was equal to 15 mm). The simplest explanation of the appearance of the electron current in the collector circuit is as follows: at the cathode the electrons have a Maxwellian velocity distribution corresponding to the cathode temperature and an average energy of the order of 0.3 to 0.4 eV. This is responsible for the electron current in the collector circuit. However, simple measurements have shown that this is not the case. The present author has used the collector as a probe and by applying to it a retarding potential relative to the cathode has determined the collector current. Fig. 3 shows the collector current I_3 as a function of the retarding

Card 4/8

21653

S/109/61/006/003/008/018

E032/E314

Electron Currents in

potential U_3 (Curve 1) for $U_a = 440$ V, $H = 51$ Oe and $P = 2 \times 10^{-5}$ mm Hg. Curve 2 in this figure gives the electron current at the collector obtained by introducing a correction for the ion current. Fig. 4 shows the logarithm of this current as a function of the retarding potential. The "temperature" of the electrons can be obtained from the slope of the straight-line portion of this curve and turned out to be equal to 253 000 °K, which corresponds to an average electron energy of 33 eV. Thus, the average energy of the electrons at the collector surface is one hundred times greater than at the cathode surface. It is clear that when $H = H_c$, the electrons somehow receive additional energy, which ensures that they reach the collector maintained at a negative potential. It is stated that no satisfactory theory of this effect is available at present. The appearance of the electron current in the cathode circuit must be taken into account in the calibration of ionisation manometers.

Card 5/8

21653

S/109/61/006/003/008/018
E032/E314

Electron Currents in

There are 4 figures, 1 table and 20 references:
7 Soviet and 13 non-Soviet.

SUBMITTED: January 18, 1960

<u>Table:</u>	<u>U, V</u>	<u>H, Oe</u>	<u>H_c, Oe</u>
	440	52.5	54.4
	480	55.5	56.8
	520	57.5	59.0
	560	60.0	61.3
	600	61.5	63.5

Card 6/8

POLYAK, I.M.

3

Sensitivity of Sulphur-Silver Photoelements to X-Rays and their Application for Dosing with X-Rays. I. M. POLYAK and M. N. DYACHENKO. Zh. Tech. Fiz., 1952, 22, 670-676. — Three different sulphur-silver photoelements (no details of composition given) were found to possess high sensitivity to X-rays, which did not change with time. The current from the photoelements depended on the spectral composition of the X-rays and therefore when used for dosing with X-rays they had to be differentially calibrated. For one of the photoelements, using a 1 mm. copper filter and a dose of 0.23 r./min., the photocurrent was 1.73×10^{-8} amp., while with a 0.1 mm. copper filter and a dose of 24.45 r./min., the current was 2.82×10^{-8} amp., so that in conjunction with a galvanometer of sensitivity 10^{-8} amp. per division it can be used as an X-ray dosimeter. Since the geometrical dimensions of the element are small enough, they can be used for determining the intensity distribution of X-rays in space. By using zinc sulphide fluorescent screens the sensitivity can be increased enough for a galvanometer of sensitivity 10^{-8} amp. per division to be used. Phys. Abs.

Chair Physic., Khar'kov Inst. Railroad Engng,
Phys. Lab., Ukr. X-Ray & Oncological Inst.

VAYSFEL'D, D.N. (Odessa, ploshchad' Sovetskoy Armii, d.l. kv.76); POLYAK, I.S.

Changes in the spine in chronic solaritis; abstract. Ortrop. travm.
i protez. 22 no.1:84 Ja '61. (MIFA 14:5)

1. Iz kurortnoy polikliniki "Kuyal'nik" Odesskogo kurortnogo
upravleniya (glavnyy vrach - I.I.Litinetskiy).
(SPINE) (SOLAR PLEXUS---DISEASES)

HARI, Bela; SOMOSKOI, Gabor; SOMOGYI, Miklos; POLYAK, Janos; HALASZ, Jozsef;
DONATH, Bela

Excerpts from remarks made at the 7th plenary session of the
National Council of Trade Unions. Munka 10 no.3:8 Mr '60.

1. Veszprem megyei Szakszervezeti Megyei Tanacs elnoke (for Hari)
2. Egitok Szakszervezetének fotitkara (for Somoskoi)
3. Magyar Szocialista Munkaspart Politikai Bizottsaganak tagja, a Szakszervezetek Orszagos Tanacsa elnoke (for Somogyi)
4. Vasas Szakszervezet titkara es "Munka" szerkesztobizottsaganak tagja (for Polyak)
5. MEDOSZ fotitkara (for Halasz)
6. Szakszervezetek Orszagos Tanacsa munkavedelmi osztalyanak vezetoje (for Donath).

POLYAK, Janos; SZABO, Zoltan, dr.; BESENYEI, Miklos

Forum of the workers of Hungary; questions to be discussed at
the 20th Congress of the Hungarian Trade Unions. Munka 13
no.4:1-4 Ap '63.

1. Vas- és Fémmipari Dolgozók Szakszervezetének fotitkara (for
Polyak). 2. Orvos-Egészségügyi Dolgozók Szakszervezete fotitkara
(for Szabo). 3. Postások Szakszervezetének fotitkara (for
Besenyei).

POLYAK, Janos

The Iron and Metal Industry Workers' Trade-Union for accelerating
the development of the machine industry. Munka 12 no.10:8-9
0 '62.

1. Vas- es Femipari Dolgozok Szakszervezete fotitkara.

FRANTSUZOV, B.L., kand.med.nauk, POLYAK, L.A., FEYGIN, N.P., (kiyev)

Antibiotic therapy in chronic highmoritis. Vest.otc.-rin. 20 no.4
104-105 J1-Ag '58 (MIRA 11:7)

(ANTIBIOTICS)

(SINUSITIS)

POLYAK, I.M.

Electric current in Penning tubes. Radiotekh. i elektron. 6
no.3:395-398 Mr '61. (MIRA 14:3)
(Electron tubes)

AUTHORS: Kopylov, Igor' Petrovich, Candidate of Technical Sciences, Assistant to the Chair of Electrical Machines at the Moscow Institute of Power Engineering, SOV, 161-58-1-17/33
Polyak, Leonid Moiseyevich, Engineer at the ~~Experimental Plant~~ of the Scientific Research Institute of Electrical Industry;
Radin, Vladimir Isaakovich, Engineer at the Plant imeni Vladimir Il'ich

TITLE: Electrodynamic Differential Amplifier With Bias Magnetization
(Differentsial'nyy elektromashinnyy usilitel' s podmagnichivaniyem)

PERIODICAL: Nauchnyye doklady vysshey shkoly, Elektromekhanika i avtomatika, 1958, Nr 1, pp. 136 - 142 (USSR)

ABSTRACT: The mode of operation, the design, the computation and the characteristic curves of an electrodynamical differential amplifier with bias magnetization (EDVM) is investigated. This amplifier is a combination on one aggregate of a magnetic differential amplifier with a d.c. generator. The differential circuit element in the magnetic amplifier inverses the polarity at the output of the electrodynamical amplifier, when

Card 1/3

Electrodynamic Differential Amplifier With Bias
Magnetization

SOV/ 161 -58-1-17/33

the polarity of the control signal is inversed. The EDVM warrants a high power amplification factor as high as $10^4 \div 10^6$. Such a high amplification factor is attained by the ganging of the magnetic amplifier and the d.c. motor. It considerably exceeds that of the electrodynamic transverse-field amplifier and that of the two-stage longitudinal-field amplifier. The time constant of the EDVM at a frequency of 50 c is almost equal to that of the transverse-field amplifier. It can, however, be reduced by a feed-back and by a frequency increase of the voltage feeding the input cascade. The essential advantage of the EDVM is its high stability. This is achieved because the output voltage in the whole operational range is a function of the difference of two exciting fluxes, and by a closing of a strong alternating flux through the stator, thus re-magnetizing the whole steel frame of the EDVM. Experiments showed that this new amplifier can be used in systems of automatic control. The data of the test equipment of the EDVM are given. There are 7 figures and 4 references, which are Soviet.

Card 2/3

Electrodynamic Differential Amplifier With Bias
Magnetization

SOV/ 161-58-1-17/33

ASSOCIATION: ~~Kafedra~~ elektricheskikh mashin Moskovskogo
energeticheskogo instituta (The Chair of
Electrical Machines at the Moscow Institute of Power Engineering)

SUBMITTED: January 4, 1958

Card 3/3

POLYAK, L.A.

Case of primary multiple cancer. Zhur. ush., nos. i gorl. bol. 20
no.1:72-73 Ja-F '60. (MIRA 14:5)

1. Iz otorinolaringologicheskogo otdeleniya (nachal'nik - kand.
meditsinskikh nauk B.L.Frantsuzov) Kiyevskogo okruzhnogo gosptalya.
(CANCER)

POLYAK, L.A.

Closed injury of the larynx with complete avulsion of the trachea.
Zhur. ush., nos. i gorl. bol. 21 no.1:73-75 Ja-7 '61.

(MIRA 14:6)

1. Iz otolaringologicheskogo otodeleniya (nachal'nik - kand.med.
nauk B.L. Frantsuzov) Kiyevskogo okruzhnogo gosptalya.
(LARYNX--WOUNDS AND INJURIES)
(TRACHEA--WOUNDS AND INJURIES)

KOPYLOV, I.P.; POLYAK, L.M.; RADIN, V.I.

Differential rotating amplifier with superposed magnetization.
Nauch.dokl.vys.shkoly; elektromekh. i avtom. no.1:136-142 '58.
(Rotating amplifiers) (MIRA 11:11)

KOSTRIKOV, V.S., kand.med.nauk, POLYAK, L.S.

"Diafixation" by K. Pap. Reviewed by V.S. Kostrikov, L.S. Poliak.
Ortop. travm. i protez. 19 no.3:70 My-Je '58 (MIRA 11:7)
(FRACTURES)
(PAP, K.)

BEREZKIN, V.G.; MYSAK, A. Ye.; POLYAK, L.S.

Gas-chromatographic determination of water traces in hydrocarbons. Neftekhimiia 4 no.1:156-159 Ja-F'64 (MIRA 17:6)

1. Institut neftekhimicheskogo sinteza AN SSSR imeni A.V. Topchiyeva.

POLYAK, L.V., Cand Med Sci -- (diss) "Variation in
higher nervous activity and phagocytic indicators
in the course of scarlet fever in children." Mos,
1958, 16 pp (Inst of Higher Nervous Activity of Acad
Sci USSR) 120 copies (KL, 23-56, 112)

- 146 -

POLYAK, L.V.

Changes in the higher nervous activity and phagocyte index in
scarlet fever in children [with summary in English]. *Pediatrics*
36 no.4:65-70 Ap'58 (MIRA 11:5)

1. Iz infektsionnogo otdela (zav. - prof. B.G. Shirvindt) Nauchno-
issledovatel'skogo pediatricheskogo instituta Ministerstva zdramo-
okhraneniya RSFSR (dir. - kand.med.nauk B.N. Karachevtseva).
(SCARLET FEVER)

POLYAK, L. V.

Effect of the drug "Regenerator" on the course of scarlet fever and complicating diseases. Vop.okh.mat. i det. 1 no.5:91-93 S-0 '56.
(MIRA 9:11)

1. Iz Gosudarstvennogo nauchno-issledovatel'skogo pediatricheskogo instituta Ministerstva zravookhraneniya RSFSR (dir. - kandidat meditsinskikh nauk V.N.Kzrachevtseva), Moskva.
(SCARLET FEVER) (PHARMACOLOGY)

USSR/Human and Animal Physiology - The Nervous System.

T

Abs Jour : Ref Zhur Biol., No 3, 1959, 13249

Author : Polyak, L.V.

Inst : ~~USSR Academy of Sciences~~

Title : Changes in Higher Nervous Activity and Phagocytic
Index in Children with Scarlatina

Orig Pub : Pediatriya, 1958, No 4, 65-70

Abstract : No abstract.

Card 1/1

AVANESOVA, A.G.; RYABINSKAYA, T.F.; POLYAK, L.V.

Mycerin therapy in colienteritis and other infectious gastrointestinal diseases in young children. Sov.med. 25 no.6:105-109 Je '61.
(MIRA 15:1)

1. Iz kliniki detskikh infektsiy (zav. - prof. L.D.Lebedev) II
Moskovskogo meditsinskogo instituta imeni N.I.Pirogova (dir. - dotsent
M.G.Sirotkina) i iz infektsionnogo otdela (zav. - prof. B.G.Shirvindt)
Gosudarstvennogo nauchno-issledovatel'skogo pediatricheskogo instituta
(dir. - doktor meditsinskikh nauk A.P.Chernikova) RSFSR.
(ANTIBIOTICS) (GASTROENTEROLOGY)

POLYAK, L.V., kand.med.nauk

Clinical aspects of salmonellosis in children. *Pediatrics* 38
no.423-7 Apr. '60. (MIRA 16:7)

1. Iz Nauchno-issledovatel'skogo pediatricheskogo instituta
Ministerstva zdravookhraneniya RSFSR (dir. A.P.Chernikova)
(INTESTINES--DISEASES) (SALMONELLA)

SAVINA, M.F.; POLYAK, L.V.; RYABINSKAYA, T.F.

Experience in the work of the diagnostic children's gastroin-
testinal department of the Fourth Municipal Clinical Hospital.
Nauch. trudy Ghetv. Mosk. gor. klin. bol'. no. 1:40-46 '61. (MIRA 16:2)

1. Otdel ostrykh detskikh infektsiy Gosudarstvennogo nauchno-
issledovatel'skogo instituta Ministerstva zdравookhraneniya
RSFSR (zav. otdelom prof. B.G. Shirvindt, direktor - doktor med.
nauk A.P. Chernikova) i Moskovskaya gorodskaya klinicheskaya
bol'nitsa No.4 (glavnyy vrach - G.F. Papko).
(ALIMENTARY CANAL-DISEASES) (MOSCOW-CHILDREN-HOSPITALS)

POLYAK, L.V.

Problem of the clinical aspects of salmonellosis in children.
Pediatria 38 no.1:3-6 '60. (MIRA 13:10)
(SALMONELLA)

POLYAK, Lidiya Vasil'yevna. kand. med. nauk; POTAPOVA, I.N. red.;
BASHMAKOV, G.M. tekhn. red.

[Scarlet fever in children]Skarlatina u detei. Moskva, Med-
giz, 1962. 14 p. (MIRA 16:1)
(SCARLET FEVER)

100

POTENTIOMETRIC DETERMINATION OF IRON AND COPPER IN ALUMINUM ALLOYS OF THE DURALUMIN TYPE. I. Ya. Polyak, *Zashchita Lab.* 12, 975 X(1946) (in Russian). To det. Fe and Cu in an alloy, dissolve 1.0 g. in a mixt. of 1 part concd. HNO_3 , 3 parts concd. HCl , and 6 parts concd. H_2SO_4 . Evap. to fumes and dil. with 30-40 ml. of 0.5 N H_2SO_4 . Filter if necessary, introduce a stream of CO_2 for about 2 min., and add 1.5 ml. of CrSO_4 soln. Then oxidize with $\text{K}_2\text{Cr}_2\text{O}_7$ soln. In an atom. of CO_2 , titrate potentiometrically with a standardized soln. of CeSO_4 . After the first break in the titration curve, corresponding to the reduction of Cr^{6+} , measure the vol. of reagent required to reduce Fe^{3+} to Fe^{2+} . Then add 0.7-0.8 g. of KBr and titrate further to reduce Cu^{2+} to Cu^+ . Standardize the CrSO_4 soln. against $\text{Fe}_2(\text{SO}_4)_3$ soln. and against CuSO_4 potentiometrically. The results are satisfactory. R. I. C.

6

ASB-514 METALLURGICAL LITERATURE CLASSIFICATION

11

11

***Potentiometric Methods for Determining Aluminium in Electron-Type Magnesium Alloys.** I. Ya. Polyak (Zavod. Lab., 1946, 12, (3), 268-275). (In Russian). A method is described for determining Al by titration with NaF and another, very rapid, method by titration with NaOH. N. A. ...

ASAC-514 METALLURGICAL LITERATURE CLASSIFICATION

11

m

***Potentiometric Determination of Iron and Copper in Duralumin-Type Alloys.** L. Ya. Polyak (*Zavod. Lab.*, 1946, 12, (11-12), 975-978).—[In Russian]. A description of the method.—N. A.

ASD SCA METALLURGICAL LITERATURE CLASSIFICATION

POLYAK, L. Ya.

180770

USSR/Metals - Analysis, Aluminum Alloys

Nov 50

"Potentiometric Determination of Zinc in Aluminum Alloys," L. Ya. Polyak

"Zavod Lab" No 11, pp 1299-1301

Possibility of binding large amt of aluminum into oxalate complex stable under conditions of zinc pptn has been established, thus making possible potentiometric detn of zinc in presence of aluminum as base of alloy. Two variations were developed: 1st variation based on detn of zinc in presence of aluminum only and 2d--in the presence of all, except copper, components of aluminum alloy. Accuracy of detn is $\pm 0.1\%$.

180770

IOLYAK, L. YA.

IA 159T59

USSR/Metals - Alloys, Magnesium
Potentiometers Jan 50

"Potentiometric Method for Determination of
Zinc in Magnesium Alloys," L. Ya. Polyak,
F. M. Shemyakin, 5 pp

"Zavod Lab" Vol XVI, No 1

Demonstrates possibility of potentiometric de-
termination of zinc with potassium ferrocyanide
solution in presence of main components of mag-
nesium alloys--magnesium, manganese and aluminum--
and also of cerium and zirconium. Method is en-
tirely different from usual method based on

159T59

USSR/Metals - Alloys, Magnesium (Contd) Jan 50

separating zinc from all components of an alloy
and its subsequent titration with potassium
ferrocyanide. Develops two methods. Second
one permits determination of small quantities
of zinc. Analysis time is 40 min; accuracy
 $\pm 0.02-0.03\%$.

159T59

CA

Potentiometric determination of zinc in aluminum alloys.
L. Ya. Polyak. *Zhurnal Khim. 10, 1290-1301(1950).*—
Since large quantities of Al can be bound as oxalate complex which is stable to $K_4Fe(CN)_6$ under conditions of pptn. of Zn, it is possible to det. Zn potentiometrically in Al or Mg alloys. Two variations are given. Boil the soln. of the sample in 20% KOH to remove all H bubbles, dil. with hot H_2O to 150 ml., filter (3% hot KOH wash), and add 9 N H_2SO_4 to the hot filtrate until pptd. hydroxides dissolve, then 5 ml. in excess, add 50-60 ml. of satd. oxalic acid soln. and evap. to 150 ml. Heat to 75°, and titrate Zn potentiometrically with $K_4Fe(CN)_6$, after adding 15 drops dil. $K_4Fe(CN)_6$ (0.5 g. per l.). In the 2nd method dissolve the sample in hot 6 N H_2SO_4 , filter off the insol. Cu, treat the filtrate at the b.p. with 0.05 N $KMnO_4$ until a stable pink color remains, add 50-60 ml. satd. oxalic acid and 20 ml. of satd. K_2SO_4 soln., heat to 75°, and titrate as above. G. M. Kosolapoff

POLYAK, L.YA.

Chemical Abstracts
Vol. 48 No. 5
Mar. 10, 1954
General and Physical Chemistry

1 The rate of reactions occurring in potentiometric titration. L. Ya. Polyak and B. N. Kabanov. *Zhur. Anal. Khim.* 8, 243-25 (1953). The rate of the electrochem. reaction $\text{Fe}(\text{CN})_6^{4-} \rightarrow \text{Fe}(\text{CN})_6^{3-}$ on a Pt electrode was studied as it is affected by the concn. of the salts themselves and by other electrolytes. Detns. were made at a polarizing current of 10^{-4} - 10^{-2} amp./sq. cm. Following the immersion of the electrode in a soln. the equil. potential was noted., and then increasingly higher currents were passed. Curves were obtained for cathodic and anodic polarization. One series of curves was obtained for solns. contg. 1×10^{-2} - $1 \times 10^{-4} M$ ferricyanide and $1 \times 10^{-4} M$ ferrocyanide. Another set of curves was obtained for 2×10^{-4} and $2 \times 10^{-6} M$ ferricyanide and $2 \times 10^{-4} M$ ferrocyanide. Curves were obtained in solns. 0.05 - $1 M$ with respect to H_2SO_4 , in $1 M$ H_2SO_4 - $0.2 M$ K, in $0.2 M$ K, and in solns. free of both H_2SO_4 and K. Curves were also obtained for solns. $0.5 M$ and $0.54 M$ with respect to Al and for solns. contg. the same concns. of Al but in which the Al was fixed as oxalate. Calens. based on these curves show that in the system $\text{Fe}(\text{CN})_6^{4-} \rightarrow \text{Fe}(\text{CN})_6^{3-}$ the potential is established within seconds and that the slowness of potentiometric titration is attributable rather to the slowness of chem. reactions taking place in the body of the soln. The titration process was accelerated in the presence of H and K. Contrary to opinion found in the literature, Al salts do not affect the Pt electrode reducing its sensitivity. Al apparently reacts with $\text{Fe}(\text{CN})_6^{4-}$, forming sol. but nondissoeg. complexes. Fixing Al as oxalate removed this difficulty. In strongly acid solns. the reaction taking place is $2\text{H}^+ + \text{Fe}(\text{CN})_6^{4-} + e \rightarrow \text{H}_2\text{Fe}(\text{CN})_6^{3-}$ rather than $\text{Fe}(\text{CN})_6^{4-} + e \rightarrow \text{Fe}(\text{CN})_6^{3-}$. Consequently the electrode potential is given by $e = e_0 + (RT/F) \ln \{[\text{Fe}(\text{CN})_6^{3-}][\text{H}^+]^2 / [\text{H}_2\text{Fe}(\text{CN})_6^{3-}]\}$. M. Hosh

✓1181. Rate of processes occurring during potentiometric titration. I. Ya. Polyak and B. N. Kabanov
(J. Anal. Chem., USSR, 1959, 8 (5), 253-255).
The kinetics of the establishment of oxidation-reduction potentials in the system $\text{Fe(CN)}_6^{4-} \rightleftharpoons \text{H}_2\text{Fe(CN)}_6^{2-}$ on a platinum electrode are studied from polarization curves and from determinations of change of potential with time. The rate of the titration is governed by the slow rate of chemical reaction in the soln, and not by the rate of establishment of potential, which is rapid. In strongly acid soln, the reaction is—
 $2\text{H}^+ + \text{Fe(CN)}_6^{4-} + e \rightarrow \text{H}_2\text{Fe(CN)}_6^{2-}$, and not
 $\text{Fe(CN)}_6^{4-} + e \rightarrow \text{Fe(CN)}_6^{3-}$.
The electrode potential is determined by the equation—
$$e = e^\circ + \frac{RT}{F} \log \frac{[\text{Fe(CN)}_6^{4-}][\text{H}^+]^2}{[\text{H}_2\text{Fe(CN)}_6^{2-}]}$$

The reaction is accelerated by H^+ and K^+ . Al^{3+} does not render the platinum electrode insensitive, as is usually supposed, but interferes with the titration through formation of only slightly associated Al ferrocyanide.
G. S. SMITH

POLYAK, L. Ya.

✓ Analysis of electrolyte baths by the potentiometric method with Trilon B. L. Ya. Polyak. *Zarodskaya Lab.* 21, 1300-1 (1955).—Ni bath electrolyte can be analyzed for Ni (within 0.2 mg.), Cu bath electrolyte can be analyzed for Cu (usually within 0.3 g./l.), Zn and Cd electrolytes for these metal ions (within 0.4 g./l.), by treatment of the neutralized soln. with NH_4OAc , followed by excess of 0.1N Trilon B, with back titration of the Trilon B with FeCl_3 potentiometrically. The Pt-calomel electrode pair was employed. O. M. K.

CH

200

AA

22

POBYAK, L. YA.

8
4e4j

3006. The potentiometric determination of lan-
thanum, cerium, praseodymium, neodymium, and
samarium. L. Ya. Polvak and E. M. Shemyakin.
Trudy Komiiss. Anal. Khim., Akad. Nauk SSSR,
1956, 7 (10), 276-288; Ref. Zhur., Khim., 1957,
Abstr. No. 12,091. From a study of the pptn.
reactions of Ce and La with $K_4Fe(CN)_6$ and Na
oxalate, it is shown that they are hydrolytic pptn.
reactions, and that therefore in the potentiometric
titration it is possible to use any type of hydrogen
electrode (quinhydrone, antimony, glass). The
possibility is indicated of the potentiometric
determination of Ce, La, Nd, Pr and Sm with
NaOH soln. The influence of elements of the third
analytical group and Mg on the potentiometric
titration of Ce has been investigated.

C. D. KOPERN

PM
mt ja

POLYAK, L. YA.

7
9
2
Titration with Trilon B. B. N. Kabanov and L. Ya. Polyak. Zhur. anal. Khim. 11, 678-85 (1950). A no. of metals were tested as electrodes in potentiometric noncompensating titration with Trilon B in ammoniacal solns. Only W and Mo were suitable. In pairs with Pt these metals acted as indicating electrodes while Pt acted as comparison electrode. Ba, Ca and Mg form stable complexes with Trilon B at pH 7 and the behavior of these electrodes in their titration was studied. BaCl₂ solns. (0.01 N) were titrated with 0.01 N Trilon B solns. and the excess of the latter was back titrated with BaCl₂. The electrodes used were Pt-W, W-calomel, and Pt-calomel. In the titration the Pt electrode potential remained const. while the potential of W shifted at the equiv. point to the neg. side. A study of this behavior lead to the conclusion that the W electrode polarized anodically. Thus, Trilon B affected only the anodic process. It was explained that Trilon B was adsorbed on W displacing O which in turn activated the electrode causing a shift to the neg. side. In the case of Mo, Trilon B depressed the overvoltage of the anodic process and did not affect the cathodic process of O₂ ionization.

M. Hosh

RM

POLYAK, D. YA.

2891. Electrochemical behaviour of electrodes in potentiometric titrations with Trilon B (EDTA, disodium salt). D. N. Kobayev and L. Ya. Polyak. *Zhur. Anal. Khim.*, 1960, 11 (6), 678-685. — With Pt as reference electrode and W as indicator electrode the latter is polarised anodically during titration with EDTA in ammoniacal soln. It is supposed that EDTA molecules adsorbed on the surface of the electrode dislodge O and cause thereby the potential to become more negative. With Mo as indicator electrode, EDTA lowers the overvoltage for the anodic oxidation of Mo, but it has no effect on the cathodic process of ionisation of O.

G. S. SMITH

POLYAK, I. Y. I.

390. Potentiometric determination of small amounts of barium in nickel-base alloys by means of EDTA (disodium salt). I. Ya. Polyak. Zhur. Anal. Khim., 1957, 12 (2), 224-229. The sample (1 g) is dissolved in HCl and the soln. is evaporated to 20 ml, then diluted to between 250 and 300 ml, kept hot for 1-5 to 2 hr. (most of the W, Nb and Ti is pptd.) and filtered. The filtrate, mixed with 10 ml of HCl and diluted to 300 ml, is treated at the b.p. with 5 ml of dil. H_2SO_4 (1:5), and the $BaSO_4$ is filtered off on the following day. The ppt. is treated on the filter with 0.1 N EDTA (disodium salt). The $BaSO_4$ does not react with the EDTA, but impurities of Ni, Cr, Al, Mn and Fe dissolve. The residue is ignited at 700° and then dissolved in 20 ml of an ammoniacal 0.01 N soln. of the EDTA. After sufficient time for the dissolution, the excess of EDTA is titrated potentiometrically with 0.01 N $BaCl_2$ with platinum-tungsten electrodes. Insoluble SiO_2 , etc., is not filtered off as it does not interfere with the titration. The method is applicable to the determination of 0.05 to 0.5% of Ba.

G. S. SMITH

NS 11

AUTHORS: Kabanov, B. N., Polyak, L. Ya. SOV/75-13-5-6/24

TITLE: Electrochemical Behavior of Aluminum Electrode in the Process of Titration of an Aluminum Ion With Fluoride
(Elektrokhimicheskoye povedeniye alyuminiyevogo elektroda pri titrovanii iona alyuminiya ftoridom)

PERIODICAL: Zhurnal analiticheskoy khimii, 1958, Vol 13, Nr 5, pp 538-544 (USSR)

ABSTRACT: Since 1948 also pairs of aluminum nichrome electrodes have been used as electrodes in the titration of aluminum ions with a solution of sodium fluoride (Refs 2-4). In these practical applications one had neither clarified the way these pairs of electrodes operate, the cause of the displacement in potential at the end point of titration, nor the influence of the cations and anions in the solution on the electrochemical behavior of the electrodes. These questions are the subject of the treatise under review. It was discovered that the potential displacement at the end point of titration is caused by a distinct change of the stationary potential of the aluminum electrode (indicator electrode); the potential of the nichrome electrode does not change during titration. In order to clarify the character

Card 1/3

S07/75-13-5-6/24

Electrochemical Behavior of Aluminum Electrode in the Process of Titration of an Aluminum Ion With Fluoride

of the anodic and cathodic processes which determine the stationary potential of the aluminum electrode and its change in titration, polarization curves of the aluminum electrode in solutions of fluoride ions and in the presence of the ions Cl^- , CH_3COO^- , and SO_4^{2-} were established. These foreign ions were added in varying combinations and concentrations that correspond to the conditions in the titration of aluminum salts with fluorides. It turned out that the distinct potential displacement in the end point of titration in the direction of negative values is caused by the accelerated effect of the fluoride ions upon the anodic process of decomposing the aluminum (activation of the aluminum electrode). The fluoride ions have practically no influence upon the cathodic process (separation of hydrogen). Chlorine ions remove the entire passivation of aluminum by increasing the potential scope in which the fluoride ions are active. Acetate ions increase the sensitivity of the aluminum electrode toward fluoride ions. The cause for the potential displacement of the aluminum electrode

Card 2/3

SOV 75-13-6-6/24
Electrochemical Behavior of Aluminum Electrode in the Process of Titration
of an Aluminum Ion With Fluoride

at the end point of titration under the influence of fluoride ions is the same both in compensating as well as in non-compensating methods of titration. The limits of applicability of the non-compensating methods were established. They depend on the field of current density in which the fluoride ions have their activating influence on the aluminum electrode. In an experimental part the tests which were carried out are described. There are 7 figures and 6 references, 5 of which are Soviet.

ASSOCIATION: All-Union Scientific Research Institute of Aviation Materials

SUBMITTED: August 31, 1956

Card 3/3

ACC NR: AP6036389

SOURCE CODE: UR/0032/66/032/011/1317/1318

AUTHOR: Polyak, L. Ya.

ORG: none

TITLE: Photometric methods for determining zirconium and titanium in molybdenum alloys

SOURCE: Zavodskaya laboratoriya, v. 32, no. 11, 1966, 1317-1318

TOPIC TAGS: photometric analysis, zirconium, titanium, molybdenum containing alloy

ABSTRACT: The article describes a newly developed method for determining $> 0.005\%$ zirconium and $> 0.01\%$ titanium from the same weighed portion of a molybdenum alloy without separating the components of the alloy. Zirconium is determined by reaction with arsenazo III. A figure shows curves for the light absorption of arsenazo III and its complex compounds with zirconium. It is found that the optimum acidity for photometric determination is 2-6 N with respect to hydrochloric acid. With an increase of the acidity even up to 9 N, the optical density of the solutions decreases slightly. The complex compound of zirconium with arsenazo III is formed instantaneously and remains stable for a period of several days. Tungsten and titanium in the amount of from 1 to 10% in the alloy do not interfere with the determination of zirconium. Niobium reacts with arsenazo III with the formation of colored complexes. To eliminate the influence of niobium, it is bound with tartaric acid. Up to 2% niobium

Card 1/2

UDC: 543.4:535.24

L 29570-66 EWT(m)/EWT(m) LJP(c) JD/JG
ACC NR: AP6019492

SOURCE CODE: UR/0075/66/021/006/0682/0687

AUTHOR: Polyak, L. Ya.; Bashkistrova, I. S.

ORG: none

TITLE: Titrimetric and photometric determination of molybdenum in niobium alloys by means of Complexon III

SOURCE: Zhurnal analiticheskoy khimii, v. 21, no. 6, 1966, 682-687

TOPIC TAGS: molybdenum determination, molybdenum niobium alloy, titration, photometry, Complexon III

ABSTRACT: Two methods have been developed for determining large amounts of Mo(5—50%) in Mo-Nb alloys. The methods are based on the capacity of pentavalent Mo to form stable yellow compounds with Complexon III [Sodium salt of EDTA]. The titrimetric method involves addition of an excess of Complexon III to an Mo-Nb alloy solution and back titration of Complexon III in an acetate medium. The titrant is cupric sulfate; the indicator is 1-(2-pyridylazo)-2-naphthol. The second method is photometric determination of Mo in the form of its compound with Complexon III. Niobium and other components of niobium alloys are masked with tartaric acid and sodium fluoride. The two methods yield reproducible results. The procedures are described in the source. Orig. art. has: 2 figures and 1 table.

SUB CODE: 07/ SUBM DATE: 09Oct64/ ORIG REF: 002/ OTH REF: 004/ ATD PRESS: 5815
Card 1/1 CC UDC: 543.70

POLYAK, L. Ya.

Photometric determination of titanium in niobium-based alloys
by means of diantipyrylmethane. Zhur. anal. khim. 19 no.12:
1468-1470 '64 (MIRA 18:1)

POLYAK, L.Ya.; BASHEIKOVA, I.S.

Photometric determination of zirconium in magnesium and aluminum
alloys using quercetin. Zhur. anal. khim. 19 no.7:842-846 '64.
(MIRA 17:11)

POLYAK, L.Ya.

Photometric determination of titanium in aluminum and molybdenum-based alloys by means of diantipyrylmethane. Zhur.anal.khim. 18 no.8:956-960 Ag '63. (MIRA 16:12)

POLYAK, L.Ya.

Photometric determination of titanium in nickel - and iron-based
refractory alloys with the use of diantipyrylmethane. 2hur.anal.-
khim. 17 no.2:206-211 Mr-Ap '62. (MIRA 15:4)
(Titanium--Analysis) (Alloys)

POLYAK, L.Ya.

Photocolorimetric determination of calcium in magnesium
alloys with the arsenazo reagent. Zav, lab. 27 no.7:803-806
'61. (MIHA 14:7)
(Calcium--Analysis) (Magnesium alloys)

POLYAK, L.Ya.

Colorimetric reaction for iron with diantipyrylmethane. Zav. lab.
27 no. 4:388-389 '61. (MIRA 14:4)
(Iron---Analysis)

FOI/AF, L. Ia.

L. Ia. Poliak. Potentiometric determination of zinc in aluminum alloys. P. 1899

SO: Industrial Laboratory (USSR) 16, No. 11 (Nov. 1950)

POLYAK, L. Ya.

5(2)

PHASE I BOOK EXPLOITATION

SOV/3224

Mukhina, Zinaida Stepanovna, Yekaterina Ivanovna Nikitina, Lidiya Mitrofanovna Budanova, Raisa Samuilovna Volodarskaya, Lyudmila Yakovlevna Polyak, and Anna Aleksandrovna Tikhonova

Metody analiza metallov i splavov (Methods of Analysis of Metals and Alloys) Moscow, Oborongiz, 1959. 527 p. Errata slip inserted. 8,050 copies printed.

Ed. of Publishing House: T. M. Kunyavskaya; Tech. Ed.: V. P. Rozhin.

PURPOSE: This book is intended for laboratory technicians of plants and may also be of use to personnel of chemical and analytic laboratories of scientific institutions and schools of higher education.

COVERAGE: The book reviews various methods of analyzing steel, cast iron, complex iron, chromium-, nickel- and cobalt-base alloys. It also reviews methods of determining the content of elements in aluminum, magnesium and copper alloys as well as in various binary alloys. Principles of physical and chemical analysis for

Card 1/14

Methods of Analysis of Metals and Alloys

SOV/3224

metallurgical studies are briefly explained, and laboratory equipment used for this kind of analysis is described and illustrated. Methods of analysis are grouped according to the type of alloy being analyzed. Each method is described and its accuracy, theoretical basis and procedure are indicated. The application of chromatographic separation in analyzing various metal alloys is explained. The appendix contains the description of various titration solutions, the reactivation of solutions and tables indicating weights of substances used in acidimetry as well as certain oxidizers, reducing agents, conversion coefficients, atomic weights of elements, etc. V. Ye. Bukhtiarov and D. V. Romanov wrote the part entitled "Methods of Chromatographic Analysis". There are 118 references: 103 Soviet, 4 German, 3 English 2 Czech and 1 French.

TABLE OF CONTENTS:

Foreword	3
Ch. I. Principles of Physicochemical Analysis	5
Part A. Colorimetric analysis	5
Card 2/14	

Methods of Analysis of Metals and Alloys

SOV/3224

Part B. Potentiometric analysis	22
Part C. Polarographic and amperometric analysis	28
Part D. Chromatographic analysis	51
Part E. Organic reagents	53
Ch. II. Analysis of Steel and Cast Iron	62
Part A. Analysis of steel	62
General information	62
Instructions for selecting a sample for chemical analysis	62
Determination of carbon	63
Determination of sulfur	72
Determination of phosphorus	80
Determination of silicon	86
Determination of manganese	91
Determination of chromium	95
Determination of nickel	100
Determination of cobalt	105
Determination of copper	109
Determination of vanadium	114

Card 3/14

Methods of Analysis of Metals and Alloys

SOV/3224

Determination of titanium	120
Determination of tungsten	120
Determination of molybdenum	128
Determination of aluminum	133
Determination of niobium	142
Determination of tantalum	144
Determination of zirconium	146
Determination of beryllium	149
Determination of selenium	151
Determination of boron	152
Determination of tin	156
Determination of arsenic	158
Determination of lead	163
Part B. Analysis of cast iron	165
Determination of carbon content	166
Determination of silicon	166
Determination of manganese	167
Determination of sulfur	167
Determination of phosphorus	167
Determination of chromium	168
Determination of nickel	168

Card 4/14

Methods of Analysis of Metals and Alloys

SOV/3224

Determination of molybdenum	169
Determination of copper	169
Determination of titanium	170
Determination of vanadium	170
Determination of tungsten	170
Determination of magnesium	170
Determination of tellurium	174

Ch. III. Analysis of Heat Resistant Alloys	176
Part A. Determination of Major Components	176
Determination of carbon	176
Determination of manganese	177
Determination of tungsten	179
Determination of titanium	185
Determination of iron	187
Determination of vanadium	191
Determination of molybdenum	192
Determination of phosphorus	194
Determination of cobalt	201

Card 5/14

Methods of Analysis of Metals and Alloys

SOV/3224

Determination of nickel	203
Determination of boron	206
Determination of beryllium	207
Determination of niobium	209
Determination of aluminum	212
Determination of zirconium	218
Determination of calcium	219
Determination of copper	220
Determination of barium	221
Part B. Determination of admixtures	222
Determination of lead	228
Determination of bismuth	231
Determination of copper	233
Determination of cadmium	235
Determination of arsenic	237
Determination of antimony	241
Determination of tin	243
Determination of zinc	246
Ch. IV. Analysis of Ferroalloys	246
Sampling	

Card 6/14

Methods of Analysis of Metals and Alloys

SOV/3224

Analysis of ferrosilicon and metallic silicon	247
Analysis of ferrochrome and metallic chromium	252
Analysis of ferromolybdenum and metallic molybdenum	261
Analysis of ferrotitanium	269
Analysis of ferromanganese and of metallic manganese	276
Analysis of ferrovanadium	280
Analysis of ferrotungsten and metallic tungsten	284
Analysis of ferrophosphorus	287
Analysis of ferrotantalum-niobium alloys	289
Analysis of calcium-silicon	290
Ch. V. Analysis of Slags	292
Sampling	292
Part A. Analysis of ordinary slags	293
Part B. Analysis of slag containing fluorine	304
Part C. Analysis of special-type slags	306
Ch. VI. Analysis of Aluminum Alloys	312
Introduction	312

Card 7/14

Methods of Analysis of Metals and Alloys

SOV/3224

Part A. Determination of major components	313
Determination of copper	313
Determination of magnesium	321
Determination of manganese	331
Determination of iron	333
Determination of silicon	336
Determination of zinc	343
Determination of calcium	351
Determination of chromium	352
Determination of nickel	354
Determination of titanium	358
Determination of vanadium	361
Determination of molybdenum	361
Determination of beryllium	362
Determination of boron	367
Determination of zirconium	369
Determination of cerium and of other rare earth elements	371
Determination of silver	373
Determination of sodium	374
Determination of aluminic acid	375
Part B. Determination of admixtures	378

Card 8/14

Methods of Analysis of Metals and Alloys

SOV/3224

Determination of tin	378
Determination of arsenic	381
Determination of antimony	384
Determination of lead	388

Ch. VII. Analysis of Magnesium Alloys

Introduction

Determination of aluminum	391
Determination of zinc	391
Determination of cadmium	391
Determination of copper	399
Determination of iron	403
Determination of silicon	408
Determination of manganese	411
Determination of nickel	413
Determination of zirconium	419
Determination of calcium	421
Determination of strontium	423

Card 9/14

Methods of Analysis of Metals and Alloys

SOV/3224

Determination of barium	428
Determination of cerium	429
Determination of bismuth	434
Determination of antimony	435
Determination of silver	436
Determination of lithium	438
Determination of beryllium	439
Determination of boron	441
Determination of chlorine	441
Determination of potassium	443
Determination of sodium	445
Determination of carbon	446
Analysis of mixed metal	447
Analysis of zirconium dioxide	448

Ch. VIII. Analysis of Copper Alloys

450

Introduction

450

Part A. Analysis of bronze of BrAZh, BrAZhMts, and BrAZhN
manufacture

450

Determination of copper

451

Determination of aluminum

454

Card 10/14

Methods of Analysis of Metals and Alloys

SOV/3224

Determination of iron	457
Determination of nickel	460
Determination of bismuth	460
Part B. Analysis of tin bronze and tin-lead bronze	461
Determination of tin	462
Determination of lead	464
Determination of phosphorus	465
Determination of iron	466
Determination of aluminum	467
Determination of antimony	467
Determination of bismuth	468
Part C. Analysis of antimonial bronze	469
Determination of antimony	469
Determination of copper, nickel and zinc from the same batch	473
Determination of bismuth	475
Part D. Analysis of beryllium bronze	476
Determination of beryllium	476
Determination of nickel	479

Card 11/14

Methods of Analysis of Metals and Alloys

SOV/3224

Part E. Analysis of chromous bronze	479
Determination of chromium	479
Determination of silicon	480
Determination of copper	480
Determination of aluminum	480
Determination of iron	481
Determination of magnesium	482
Part F. Analysis of different types of brass and TsAM10-5 and TsAM9-1.5 alloys	483
Determination of iron and lead from the same batch	484
Determination of phosphorus	484
Determination of arsenic	485
Determination of bismuth	486
Determination of zinc	486
Part G. Analysis of silver solder	488
Determination of silver	488
Determination of copper	489
Determination of nickel	489
Determination of zinc and cadmium	490
Part H. Analysis of binary alloys	491
Copper-manganese alloy	491

Card 12/14

Methods of Analysis of Metals and Alloys

SOV/3224

Copper-phosphorus alloy	491
Copper-cadmium alloy	492
Copper-cobalt alloy	493
Copper-bismuth alloy	493
Lead-bismuth alloy	494
Copper-silicon alloy	494
Part I. Application of chromatographic separation in analyzing copper alloys	496
Bronze of BrAZH manufacture	496
Beryllium bronze	500
Copper-iron alloy	501
Appendix I. Titration Solution	503
Appendix II. Equivalent Weight of Substances Most Frequently Used in Acidimetry	511
Appendix III. Equivalent Weights of Most Frequently Used Oxidizers	511
Card 13/14	

SOV/137-57-1-1625

Translation from: Referativnyy zhurnal. Metallurgiya, 1957, Nr 1, p 216 (USSR)

AUTHORS: Polyak, L. Ya., Shemyakin, F. M.

TITLE: On the Potentiometric Determination of Lanthanum, Cerium, Praseodymium, Neodymium, and Samarium (K voprosu o potentsiometricheskom opredelenii lantana, tseriya, prazeodima, neodima i samariya)

PERIODICAL: Tr. Komis. po analit. khimii, AN SSSR, 1956, Vol 7 (10), pp 276-288

ABSTRACT: It is established that Ce and La form $\text{MeKFe}(\text{CN})_6$ -type compounds with $[\text{Fe}(\text{CN})_6]^{4-}$ and $\text{Me}_2(\text{C}_2\text{O}_4)_3$ with $[\text{C}_2\text{O}_4]^{2-}$, which can be determined by potentiometric titration using quinhydrone, antimony, and glass electrodes paired with saturated cathode electrolyte. Elements that are precipitated by the above reagents or form complex compounds with them impede the determination. The accuracy of the determination is equivalent to that of the gravimetric method of precipitating rare earths in the form of oxalates.

Z. G.

Card 1/1

GORELOVA, A.A.; POIYAK, L.Ya.

Potentiometric method for determining aluminum in heat-resistant
nickel-base alloys. Zav.lab. 25 no.3:285-287 '59. (MIRA 12:4)
(Heat-resistant alloys) (Aluminum--Analysis)
(Potentiometric analysis)

SCV/32-25-6-9/53

5(2)

AUTHORS: Busev, A. I., Polyak, L. Ya.

TITLE: Determination of Cadmium in Magnesium Alloys With the Use of Nickel Diethyl Dithiophosphate (Opredeleniye kadmiya v magniyevykh splavakh s primeneniye dietilditiofosfata nikelya)

PERIODICAL: Zavodskaya Laboratoriya, 1959, Vol 25, Nr 6, pp 668 - 669 (USSR)

ABSTRACT: A method was introduced, to be applied for the gravimetric as well as the volumetric determination of cadmium in magnesium alloys. Zn, Al, Ce, Zr et al may be present as well. The method is based on the precipitation of Cd with nickel diethyl dithiophosphate and the weighing of the precipitate $[(C_2H_5O)_2PSS]_2Cd$ or a titration of the precipitate with a 0.05 n iodine solution. The presence of Mg and of the abovementioned alloy elements of magnesium alloys does not disturb the determination (Table 1). The iodometric titration is based on the reaction:
 $[(C_2H_5O)_2PSS]_2Ni + J_2 = (C_2H_5O)_2PSS - SSP(OC_2H_5)_2 + NiJ_2$
 and is considerably quicker than the gravimetric determination, the analytical accuracy being the same (Table 2). The course

Card 1/2

Determination of Cadmium in Magnesium Alloys With the SOV/32-25-6-9/53
Use of Nickel Diethyl Dithiophosphate

of analysis is described. There are 2 tables and 2 Soviet
references.

Card 2/2

5(2)

AUTHORS:

Gorelova, A. A., Polyak, L. Ya.

SOV/32-23-3-9/62

TITLE:

Potentiometric Method of Determining Aluminum in Heat-resistant Alloys on a Nickel Basis (Potentsiometricheskiy metod opredeleniya alyuminiya v zharoprochnykh splavakh na nikel'evoy osnove)

PERIODICAL:

Zavodskaya Laboratoriya, 1959, Vol 25, Nr 3, pp 285 - 287 (USSR)

ABSTRACT:

S. K. Chirkov (Ref 3) suggested a potentiometric method of titration of aluminum with sodium fluoride in a hydrochloric medium buffered with sodium acetate by using an Al-Ni/Cr electrode. On the basis of this suggestion a rapid method for the determination of aluminum in heat-resistant alloys on a nickel basis was devised without the necessity of separating aluminum. The two reactions $Al^{3+} + 6 F^{-} \longrightarrow [AlF_6]^{3-}$ and $Al^{3+} + 3 F^{-} \longrightarrow AlF_3$ were investigated and two varieties of titrations were carried out (Fig). It was found that Ti(III) (0.2 - 2.0%), Ti(IV) (up to 0.8%), Cr(III) (up to 14%),

Card 1/2